

PERFORMANCE OF INTEGRATED TREATMENT SYSTEM OF PHOTO-  
FENTON AND MEMBRANE BIOREACTOR FOR SYNTHETIC SPENT  
CAUSTIC WASTEWATER

HANNA ILYANI BINTI ZULHAIMI

UNIVERSITI TEKNOLOGI MALAYSIA

PERFORMANCE OF INTEGRATED TREATMENT SYSTEM OF PHOTO-  
FENTON AND MEMBRANE BIOREACTOR FOR SYNTHETIC SPENT  
CAUSTIC WASTEWATER

HANNA ILYANI BINTI ZULHAIMI

A thesis submitted in fulfillment of the  
requirements for the award of the degree of  
Master of Engineering (Chemical)

Faculty of Chemical Engineering  
Universiti Teknologi Malaysia

MARCH 2014

Demi Agama, Bangsa dan Negara

## ACKNOWLEDGEMENTS

I am grateful to Allah S.W.T for giving me the blessing, guidance, strength, health and opportunity to complete this study.

My deepest gratitude goes to my supervisor, Assoc. Prof. Dr. Zainura Zainon Noor, who is continuously and influentially, conveyed a spirit of exploration in regard to research and scholarship. Without her persistent help and ideas this dissertation would not have been possible.

Next is my honorable Advisors, Assoc. Prof. Dr. Azmi Aris and Assoc. Prof. Dr. Mohd Ariffin Abu Hassan for his academic guidance and responsiveness to all of my research enquiries. I am extremely grateful and indebted to them for their expert, sincere and valuable guidance extended to me.

Finally, I would like to extend deep appreciation to my research team for their encouragement. Not forgetting to my parents for their unconditional love and support. I also would like to extend my gratitude to Universiti Teknologi Malaysia and Universiti Malaysia Perlis for providing financial support and facilities.

I also placed on record, my sense of gratitude to whom, directly or indirectly, lent their helping hand in this venture.

## ABSTRACT

The potential of a combined integrated treatment system using photo-Fenton process and membrane bioreactor (MBR) was studied in the treatment of spent caustic wastewater with initial COD and sulfide of 1990-2740 mg/L and 82-165 mg/L, respectively and at biodegradability level ( $BOD_5/COD$ ) of 0.39-0.52. Photo-Fenton process was carried out to reduce influent toxicity that can destroy living microorganism in the MBR. Molar ratio of hydrogen peroxide to COD values ( $[H_2O_2/COD]$ ), ferrous ion to hydrogen peroxide  $[Fe^{2+}/H_2O_2]$ , and reaction time were studied on degradation of chemicals contained in the wastewater within the range of 1.5-2.5, 0.05-0.15 and 0-60 minutes. A statistical experimental design using central composite design (CCD) was employed to describe the relationship of individual and combined parameters towards the degradation. The optimum conditions were selected based on the highest percentage of COD and sulfide removal as well as acceptable  $BOD_5/COD$  with minimum value of molar ratio  $[H_2O_2/COD]$  and  $[Fe^{2+}/H_2O_2]$  within reaction time range. Optimum conditions for molar ratio  $[H_2O_2/COD]$  and  $[Fe^{2+}/H_2O_2]$  at pH of 3 were 1.5 and 0.05 with reaction time of 47.84 minutes. The optimum condition for MBR was evaluated based on critical flux value. It was observed that critical flux was achieved at flux of 6 and 9.5  $L\ m^{-2}\ h^{-2}$ . The achievement of optimal conditions for both processes led to the assessment of the integrated treatment. It was found that the overall percentage removal of COD and sulfide were 99.27–99.92% and 100% leading to final effluent with COD and sulfide concentrations of 20 mg/L and 0.001 mg/L, respectively. It can be concluded that the integrated treatment system combining photo-Fenton and MBR could be an efficient alternative method for spent caustic wastewater.

## ABSTRAK

Potensi sistem rawatan bersepadu yang menggabungkan proses foto-Fenton dan bioreaktor membran (MBR), telah dikaji sebagai suatu kaedah rawatan air sisa kaustik dengan nilai kepekatan awal COD dan sulfida yang berjulat 1990-2740 mg / L dan 82-165 mg/L. Proses foto-Fenton telah dilaksanakan bagi mengurangkan ketoksikan influen yang boleh memudaratkan mikroorganisma yang terdapat di dalam MBR. Nisbah molar hidrogen peroksida kepada nilai COD ( $[H_2O_2/COD]$ ), ion ferus kepada hidrogen peroksida  $[Fe^{2+} / H_2O_2]$  dan masa tindakbalas telah diselidiki kesannya terhadap penguraian bahan kimia di dalam air sisa kaustik dengan julat nisbah masing-masing, iaitu 1.5-2.5, 0.05-0.15 dan 0-60 minit. Satu reka bentuk eksperimen berteraskan kaedah statistik yang digelar reka bentuk komposit berpusat (CCD), telah digunakan untuk mentafsirkan hubungan di antara pembolehubah tunggal dan gabungan terhadap penguraian air sisa kaustik. Keadaan optimum proses rawatan telah dipilih berdasarkan peratusan susutan nilai kepekatan COD dan sulfida yang tertinggi. Manakala, nilai optimum nisbah  $BOD_5/COD$  pula dikira berdasarkan kepada nisbah molar paling minimum,  $[H_2O_2/COD]$  dan  $[Fe^{2+} / H_2O_2]$  dalam kadar masa tindakbalas yang ditetapkan. Proses foto-Fenton yang optimum ialah semasa nisbah molar  $[H_2O_2/COD]$  dan  $[Fe^{2+} / H_2O_2]$  pada pH 3 adalah 1.5 dan 0.05 dengan masa tindakbalas, 47.84 minit. Selain itu, keadaan optimum bagi proses MBR pula dianggar berdasarkan nilai fluks kritikal yang diperolehi daripada ujikaji iaitu 6 dan 9.5 L m<sup>-2</sup> h<sup>-2</sup>. Hasil kajian ini mendapati, peratusan susutan keseluruhan nilai kepekatan COD dan sulfida adalah masing-masing sebanyak 99.27-99.92% dan 100% yang membawa kepada kepekatan akhir COD dan sulfida bagi efluen, masing-masing sebanyak 20 mg / L dan 0.001 mg / L. Oleh itu, dapat dirangkumkan bahawa sistem rawatan bersepadu yang menggabungkan proses foto-Fenton dan MBR mampu menjadi satu kaedah alternatif untuk merawat air sisa kaustik.

## TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENTS	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	xi
	LIST OF FIGURES	xiii
	LIST OF ABBREVIATIONS	xvi
	LIST OF SYMBOLS	xvii
	LIST OF APPENDICES	xix
<b>1</b>	<b>INTRODUCTION</b>	<b>1</b>
	1.1 Background of the Study	1
	1.2 Statement of Problems	3
	1.3 Objectives of the Study	4
	1.4 Scope of the Study	4

1.5	Significance of the Study	5
1.6	Thesis Layout	6
<b>2</b>	<b>LITERATURE REVIEW</b>	<b>8</b>
2.1	Introduction	8
2.2	General View of Caustic Solution	8
2.3	Classification of Spent Caustic	10
2.4	Effect of Spent Caustic to the Environment	11
2.5	Industrial Process Configuration	15
2.6	Treatment Technology for Spent Caustic Wastewater	16
2.6.1	Chemical Treatment	16
2.6.1.1	Advance Oxidation Process	16
2.6.1.2	Chemical Precipitation	20
2.6.2	Biological Treatment	22
2.6.3	Conclusion	23
2.7	Photo-Fenton Process for Wastewater	25
2.7.1	Overview	25
2.7.2	Mechanism of Photo-Fenton Process	25
2.8	Effect of Photo-Fenton Process	29
2.8.1	Wavelength and Power of UV Light	30
2.8.2	Initial Oxidant Concentration	31
2.8.3	Catalyst Concentration	33
2.8.4	Initial Parent Compound	34
2.8.5	pH	34



2.8.6	Temperature	36
2.9	Membrane Technology for Wastewater	37
2.9.1	Categories and Configuration of Membrane	37
2.9.2	Principle of Design of MBR	40
2.9.3	Fouling in Membrane	42
<b>3</b>	<b>METHODOLOGY</b>	<b>47</b>
3.1	Introduction	47
3.2	Experimental Framework	47
3.3	Synthetic Spent Caustic Wastewater	50
3.4	Chemicals	50
3.5	Analytical Methods Parameters	51
3.6	Photo-Fenton Study	52
3.6.1	Experimental Design	52
3.6.2	Experimental Procedures	55
3.6.3	Membrane Surface Morphology	57
3.7	Membrane Bioreactor Setup	58
3.7.1	Inoculums: Source and Acclimatize Condition	58
3.7.2	MBR Operation: Experimental Setup	59
3.7.3	Critical Flux	60
3.8	Integrated System of Photo-Fenton and Membrane Bioreactor	62
<b>4</b>	<b>RESULTS AND DISCUSSION</b>	<b>65</b>
4.1	Introduction	65

4.2	Adequacy of Fitting Model	68
4.3	Development of Empirical Model	74
4.4	Response Surface and Optimization	76
4.5	Surface Morphology of Membrane	80
4.6	Development of Biomass	83
4.7	Critical Flux Determination	85
4.8	Assessment of Integrated Treatment System	88
4.8.1	COD Removal	88
4.8.2	Sulfide Removal	89
4.8.3	Overall Assessment	91
<b>5</b>	<b>CONCLUSION AND RECOMMENDATIONS</b>	<b>94</b>
5.1	Conclusion	94
5.2	Recommendations	95
	<b>REFERENCES</b>	<b>97</b>
	Appendices A-G	111-168

## CHAPTER 1

### INTRODUCTION

#### 1.1 Background of the Study

The use of caustic solution in oil refineries and petrochemical industries to remove hydrogen sulfide and organic compounds from hydrocarbon compositions has led to the generation of large amount of spent caustic. The spent caustic wastewater are usually found in the prewashing and sweetening stages. Typically, they are categorized into two types: phenolic spent caustic and sulfidic spent caustic (Olmos *et al.*, 2004). The types of spent caustic wastewater are differentiated by their phenolic and free NaOH contents.

Most spent caustic wastewater are sent off-site to other industries that require them or are treated before disposal. For instance, in the pulp and paper industry, spent caustic was utilized in the mill's Kraft process, replacing caustic solution (Sipma *et al.*, 2004). On the other hand, spent caustic wastewater could be sent directly to the wastewater treatment plant. The conventional method of disposing spent caustic was by dumping in a deep well or ocean (Kolhatkar and Sublette, 1996; Sipma, *et al.*, 2004). This practice was not favoured because of the notorious impact of spent caustic wastewater especially on the ecosystem. One of the physico-chemical

treatment technologies that is usually employed is wet air oxidation (WAO). Generally, WAO is characterized by high temperature and pressure, with the additional high investment and maintenance costs and has safety implication.

Advanced oxidation processes have proven to be one of the most effective methods for the removal of pollutants in wastewater. Such processes include catalytic ozonation, heterogeneous photocatalysis with  $\text{TiO}_2$ , Fenton reaction and photo-Fenton reaction. Fenton and photo-Fenton have been used in many applications because of the abundance of raw materials, as well as their non-toxic nature and require low investment (Farre *et al.*, 2008). Fenton reaction involves ferrous ion and hydrogen peroxide reacting to produce hydroxyl radical ( $\bullet\text{OH}$ ) which is known to be strong in its oxidizing properties when in acidic solution. The presence of ultraviolet light, which known as photo-Fenton, will enhance the production rate of  $\bullet\text{OH}$ , hence its efficiency is increased (Malato *et al.*, 2007).

Of recent, several studies (Bani-Melhem and Elektorowicz, 2011; Farizoglu and Keskinler, 2006; Hai *et al.*, 2012) have led to the development of a new type of membrane that gives a promising tool to treat various types of wastewater (Chen *et al.*, 2009; Mohammed *et al.*, 2008; Rosenberger *et al.*, 2002). Membrane bioreactor has become an alternative to conventional biological treatment due to its many benefits. Its characteristics that utilize micro and ultra-filtration membrane substitute the conventional sedimentation so as to produce clear effluent from the sludge. Apart from its ability to produce high quality effluent (Mohammed, *et al.*, 2008; Mutamim *et al.*, 2013), the membrane, which has porous surface of  $0.2\mu\text{m}$  or less, manages to retain bacteria and viruses in the reactor (Rosenberger, *et al.*, 2002; Stephenson *et al.*, 1996).

Though there are few studies on combining biological treatment with advanced oxidation process, the integrated method that utilizes photo-Fenton and membrane bioreactor to treat spent caustic wastewater has not been reported. The goal of this study is to treat spent caustic wastewater in order to achieve maximum

percentage removal of COD and sulfide at acceptable biodegradability with minimum usage of Fenton reagents. It is believed that spent caustic wastewater could be treated properly in this way to meet with standard quality requirements and which can then be implemented in real wastewater treatment plant.

## 1.2 Statement of Problems

The negative environmental impacts of spent caustic wastewater on human health have been reported (Abdulah *et al.*, 2011; B. Kumfer, 2010; de Graaff *et al.*, 2011; de Graaff *et al.*, 2012; Felix Davila, 2007). The confirmation on the toxic effect has been outlined in Environmental Quality Act Regulation under category 4 scheduled waste SW 402 which usually contains corrosive or hazardous waste where it is commonly disposed to a licensed off-site recycler (Malaysia, 1979). This also in line with US Resources Conservation and Recovery Act that assigned spent caustic under D003 (reactive sulfide) hazardous waste (Sheu and Weng, 2001). Wet air oxidation is the existing treatment process for caustic wastewater. However, its operating condition requires high pressure and temperatures and it is one of its drawbacks in terms of safety and costs. Few studies have been done to find alternative treatment methods for spent caustic, like coagulation (Demirci *et al.*, 1998) and electrochemical process (Yan *et al.*, 2011). However, the methods are characterized by low removal of COD and very low rates of reaction which restrict their application (Demirci, *et al.*, 1998; Diya'uddeen *et al.*, 2011). Another factor is the single nature of the treatment methods studied. This study proposed an integrated system that consists of chemical treatment (photo-Fenton) and biological treatment (membrane bioreactor) which are expected to synergistically give high quality effluent when compared with single process. Furthermore, the integrated treatment will also serve as reference for further investigation with regard to the combination of Fenton and biological processes and the ability of microbes to withstand wastewater that has been chemically treated using the photo-Fenton process.

### **1.3 Objectives of the study**

The following are the objectives of the study:

- (i) To obtain optimum operating parameters (reaction time, dosage of oxidant and dosage of catalyst) for photo-Fenton process for synthetic spent caustic wastewater;
- (ii) To assess the effect of photo-Fenton optimum conditions on biomass and membrane;
- (iii) To obtain optimum operating condition for membrane bioreactor prior to integration of treatment system employing photo-Fenton and membrane bioreactor for treating synthetic spent caustic wastewater.

### **1.4 Scope of the Study**

The study was conducted within the scope specified below:

- The optimum operating condition for photo-Fenton process were evaluated based on highest percentage removal of chemical oxygen demand (COD) and sulfide removal with acceptable biodegradability level. In this study, the amounts of oxidant and catalyst as well as reaction time were chosen to be the operating conditions for photo-Fenton process. Response surface methodology (RSM) and the method of Central Composite Design (CCD) were implemented in this study using Design-Expert Software;
- The optimum operating condition for membrane bioreactor was determined using flux-stepping method. Critical flux was achieved in order to avoid

membrane biofouling where cleaning will be needed. The value of flux that is below the critical flux represent the operating condition. Besides, membrane bioreactor study was conducted at short-term operation where sludge retention time and biofoulant will not be studied and the membrane will undergo a one-time operation;

- Theoratically, photo-Fenton process will generate free radicals to degrade the organic compounds into simpler species. Although it has an advantage in degradation process, the concern on free radical compounds arises on whether they are toxic to the microbes. To solve the problem, acclimatization was employed as the source of microbial toxicity measurement. As for the effect of free radicals on membrane, investigation of surface morphology will be implemented. Scanning electron microscope (SEM) was used to provide qualitative justification;
- Based on the individual treatment, the optimum operating conditions and results obtained act as basis to the integrated system. The integrated treatment is assumed to be well-mixed and the performance was evaluated based on removal of COD, BOD and sulfide concentrations. The chosen parameters were analyzed on selected locations that are: (i) before entering the chemical treatment, (ii) after undergoing chemical treatment and (iii) after undergoing chemical and biological treatment. The selected parameters were compared with standard parameters assigned under the Environmental Quality (Industrial Effluent) Regulation 2009.

## **1.5 Significance of the Study**

This research serves as new route to treat spent caustic wastewater as it gives more benefits in comparison to the existing technology. It also contributes to the development of treatment technology for treating high strength wastewater. The system also employs in-situ treatment with use of smaller areas as compared to conventional treatment, which plays a significant role in selecting the treatment

plant. Since the system is operated at ambient conditions, it involves lower costs in terms of construction and maintenance. In addition, safety concern will be at a high priority compared with existing treatment that need to function at high temperature and pressure. The ability of the integrated system to treat spent caustic wastewater will serve as evidence in employing it in real wastewater treatment plants.

On the study of the toxic effect of the free radical on membrane, the result will give an overview and new findings for further research. The ability of the biomass to adapt with free radicals environment will be analysed. Finally, the justification of using a membrane that is chemically treated with photo-Fenton for wastewater treatment will have been established.

## **1.6 Thesis Layout**

This thesis consists of five chapters. Each of the chapters will be briefly discussed and they are interrelated to each other bounded by the scope of study.

**Chapter 1** deals with the overview of wastewater treatment and technologies that have been implemented highlighting its drawbacks provide the need of the study. Therefore, an alternative is proposed concerning the specific problem which has been addressed in the objectives of the study.

**Chapter 2** is a literature review on the characteristics of wastewater and covers physical and chemical characteristics of spent caustic wastewater. It also discuss the method have been carried out in the treatment of spent caustic wastewater. The factors that affect the process were addressed in this chapter.



**Chapter 3** describes the methods and technique chosen to be implemented in this study. Samples were taken for analysis and all the analytical methods are described in this chapter.

**Chapter 4** is the full result of this study. The parameters that contribute and give impact to the study will be discussed. Each of the result will be trasformed into simplified graph for further clarification.

**Chapter 5** is the conclusion obtained from the study. It gives an insight on how the objectives have been met. The chapter ends with suggestions that were made for future research.

## REFERENCES

- Abdelwahab, O., Amin, N. K. and El-Ashtouky, E. S. Z. (2009). Electrochemical removal of phenol from oil refinery wastewater. *Journal of Hazardous Materials*, 163(2-3), 711-716.
- Abdulah, S. H. Y. S., Hassan, M. A. A., Noor, Z. Z. and Aris, A. (2011). Optimization of photo-Fenton oxidation of sulfidic spent caustic by using response surface methodology. Proceedings of the 2011 *National Postgraduate Conference (NPC)*, 2011. 19-20 Sept. 2011. 1-7.
- Ahmad, N., Maitra, S., Dutta, B. K. and Ahmad, F. (2009). Remediation of sulfidic wastewater by catalytic oxidation with hydrogen peroxide. *Journal of Environmental Sciences*, 21(12), 1735-1740.
- Ahmadi, M., Vahabzadeh, F., Bonakdarpour, B., Mofarrah, E. and Mehranian, M. (2005). Application of the central composite design and response surface methodology to the advanced treatment of olive oil processing wastewater using Fenton's peroxidation. *Journal of Hazardous Materials*, 123(1-3), 187-195.
- APHA. (1995). *Standard methods for the examination of water and wastewater* (Vol. 19th ed.): American Public Health Association, American Water Works Association, Water Pollution Control Federation.
- Al Zarooni, M. and Elshorbagy, W. (2006). Characterization and assessment of Al Ruwais refinery wastewater. *Journal of Hazardous Materials*, 136(3), 398-405.
- Al-Halbouni, D., Traber, J., Lyko, S., Wintgens, T., Melin, T. and Tacke, D. (2008). Correlation of EPS content in activated sludge at different sludge retention times with membrane fouling phenomena. *Water Research*, 42(6-7), 1475-1488.
- Al-Malack, M. H. (2007). Performance of an immersed membrane bioreactor (IMBR). *Desalination*, 214(1-3), 112-127.

- Almeida, L. C., Garcia-Segura, S., Bocchi, N. and Brillas, E. (2011). Solar photoelectro-Fenton degradation of paracetamol using a flow plant with a Pt/air-diffusion cell coupled with a compound parabolic collector: Process optimization by response surface methodology. *Applied Catalysis B: Environmental*, 103(1-2), 21-30.
- Altas, L. and Buyukgungor, H. (2008). Sulfide removal in petroleum refinery wastewater by chemical precipitation. *Journal of Hazardous Materials*, 153(1-2), 462-469.
- Anderson, M. J. and Whitcomb, P. J. (2007). *DOE Simplified: Practical tools for effective experimentation* (2nd. Edition ed.). Boca Raton: Productivity Press.
- Alvarez, N., Gesan-Guiziu, G. and Daufin, G. (2007). The role of surface tension of re-used caustic soda on the cleaning efficiency in dairy plants. *International Dairy Journal*, 17(4), 403-411.
- Bach, A., Shemer, H. and Semiat, R. (2010). Kinetics of phenol mineralization by Fenton-like oxidation. *Desalination*, 264(3), 188-192.
- Bani-Melhem, K. and Elektorowicz, M. (2011). Performance of the submerged membrane electro-bioreactor (SMEBR) with iron electrodes for wastewater treatment and fouling reduction. *Journal of Membrane Science*, 379(1-2), 434-439.
- Bas, D. and Boyaci, I. H. (2007). Modeling and optimization I: Usability of response surface methodology. *Journal of Food Engineering*, 78(3), 836-845.
- Ben Hariz, I., Halleb, A., Adhoum, N. and Monser, L. (2013). Treatment of petroleum refinery sulfidic spent caustic wastes by electrocoagulation. *Separation and Purification Technology*, 107(0), 150-157.
- Benatti, C. u. T., Tavares, C. I. R. G. and Guedes, T. A. (2006). Optimization of Fenton's oxidation of chemical laboratory wastewaters using the response surface methodology. *Journal of Environmental Management*, 80(1), 66-74.
- Benitez, F. J., Acero, J. L., Real, F. J., Rubio, F. J. and Leal, A. I. (2001). The role of hydroxyl radicals for the decomposition of p-hydroxy phenylacetic acid in aqueous solutions. *Water Research*, 35(5), 1338-1343.
- Berube, P. (2010). Chapter 9 Membrane Bioreactors: Theory and Applications to Wastewater Reuse. In C. E. Isabel & I. S. f. Andrea (Eds.), *Sustainability Science and Engineering* (Vol. Volume 2, pp. 255-292): Elsevier.

- Bianco, B., De Michelis, I. and Vegliani, F. (2011). Fenton treatment of complex industrial wastewater: Optimization of process conditions by surface response method. *Journal of Hazardous Materials*, 186(2-3), 1733-1738.
- Bottino, A., Capannelli, G., Comite, A. and Mangano, R. (2009). Critical flux in submerged membrane bioreactors for municipal wastewater treatment. *Desalination*, 245(1-3), 748-753.
- Bouasla, C., Samar, M. E.-H. and Ismail, F. (2010). Degradation of methyl violet 6B dye by the Fenton process. *Desalination*, 254(1-3), 35-41.
- Chamarro, E., Marco, A. and Esplugas, S. (2001). Use of fenton reagent to improve organic chemical biodegradability. *Water Research*, 35(4), 1047-1051.
- Chatterjee, D. and Dasgupta, S. (2005). Visible light induced photocatalytic degradation of organic pollutants. *Journal of Photochemistry and Photobiology C: Photochemistry Reviews*, 6(2-3), 186-205.
- Chen, Q., Wu, P., Li, Y., Zhu, N. and Dang, Z. (2009). Heterogeneous photo-Fenton photodegradation of reactive brilliant orange X-GN over iron-pillared montmorillonite under visible irradiation. *Journal of Hazardous Materials*, 168(2-3), 901-908.
- Chen, Z.-B., Hu, D.-X., Ren, N.-Q., Tian, Y. and Zhang, Z.-P. (2009). Biological COD reduction and inorganic suspended solids accumulation in a pilot-scale membrane bioreactor for traditional Chinese medicine wastewater treatment. *Chemical Engineering Journal*, 155(1-2), 115-122.
- Coelho, A., Castro, A. V., Dezotti, M. r. and Sant'Anna Jr, G. L. (2006). Treatment of petroleum refinery sourwater by advanced oxidation processes. *Journal of Hazardous Materials*, 137(1), 178-184.
- Cornel P, W. M., Krause S. (2003). Investigation of oxygen transfer rates in full scale membrane bioreactors. *Water Sci Technol.*, 47(11), 313-319.
- Cote, P., Masini, M. and Mourato, D. (2004). Comparison of membrane options for water reuse and reclamation. *Desalination*, 167(0), 1-11.
- Cruz-Gonzalez, K., Torres-Lopez, O., Garcia-Leon, A. M., Brillas, E., Hernandez-Ramirez, A. and Peralta-Hernandez, J. M. (2012). Optimization of electro-Fenton/BDD process for decolorization of a model azo dye wastewater by means of response surface methodology. *Desalination*, 286(0), 63-68.

- Cui, M., Jang, M., Cho, S.-H., Elena, D. and Khim, J. (2011). Enhancement in mineralization of a number of natural refractory organic compounds by the combined process of sonolysis and ozonolysis (US/O<sub>3</sub>). *Ultrasonics Sonochemistry*, 18(3), 773-780.
- Damayanti, A., Ujang, Z., Salim, M. R. and Olsson, G. The effect of mixed liquor suspended solids (MLSS) on biofouling in a hybrid membrane bioreactor for the treatment of high concentration organic wastewater. (0273-1223).
- de Bo, I., Van Langenhove, H. and Heyman, J. (2002). Removal of dimethyl sulfide from waste air in a membrane bioreactor. *Desalination*, 148(1-3), 281-287.
- de Graaff, M., Bijmans, M. F. M., Abbas, B., Euverink, G.-J. W., Muyzer, G. and Janssen, A. J. H. (2011). Biological treatment of refinery spent caustics under halo-alkaline conditions. *Bioresource Technology*, 102(15), 7257-7264.
- de Graaff, M., Klok, J. B. M., Bijmans, M. F. M., Muyzer, G. and Janssen, A. J. H. (2012). Application of a 2-step process for the biological treatment of sulfidic spent caustics. *Water Research*, 46(3), 723-730.
- de Moraes, J. L. and Zamora, P. P. (2005). Use of advanced oxidation processes to improve the biodegradability of mature landfill leachates. *Journal of Hazardous Materials*, 123(1-3), 181-186.
- Demirci, S., Erdogan, B. and Ozcimder, R. (1998). Wastewater treatment at the petroleum refinery, Kirikkale, Turkey using some coagulants and Turkish clays as coagulant aids. *Water Research*, 32(11), 3495-3499.
- Diya'uddeen, B. H., Daud, W. M. A. W. and Abdul Aziz, A. R. (2011). Treatment technologies for petroleum refinery effluents: A review. *Process Safety and Environmental Protection*, 89(2), 95-105.
- El-Naas, M. H., Al-Zuhair, S., Al-Lobaney, A. and Makhoulf, S. (2009). Assessment of electrocoagulation for the treatment of petroleum refinery wastewater. *Journal of Environmental Management*, 91(1), 180-185.
- Farizoglu, B. and Keskinler, B. (2006). Sludge characteristics and effect of crossflow membrane filtration on membrane fouling in a jet loop membrane bioreactor (JLMBR). *Journal of Membrane Science*, 279(1-2), 578-587.
- Farre, M. J., Maldonado, M. I., Gernjak, W., Oller, I., Malato, S. and Domenech, X. (2008). Coupled solar photo-Fenton and biological treatment for the degradation of diuron and linuron herbicides at pilot scale. *Chemosphere*, 72(4), 622-629.

- Felix Davila, N. s. V. M. r., Chad Felch, Clay Maugans, Steve Olsen. (2007). Disposal of Spent Caustic at the Repsol YPF Refinery in La Pampilla, Peru. Proceedings of the 2007 *Environmental Conference*. September 24-25, 2007.
- Feng, J., Hu, X., Yue, P. L., Zhu, H. Y. and Lu, G. Q. (2003). Discoloration and mineralization of Reactive Red HE-3B by heterogeneous photo-Fenton reaction. *Water Research*, 37(15), 3776-3784.
- Field, R. W., Wu, D., Howell, J. A. and Gupta, B. B. (1995). Critical flux concept for microfiltration fouling. *Journal of Membrane Science*, 100(3), 259-272.
- Gander, M., Jefferson, B. and Judd, S. (2000). Aerobic MBRs for domestic wastewater treatment: A review with cost considerations. *Separation and Purification Technology*, 18(2), 119-130.
- Garg, A., Mishra, I. M. and Chand, S. (2007). Catalytic wet oxidation of the pretreated synthetic pulp and paper mill effluent under moderate conditions. *Chemosphere*, 66(9), 1799-1805.
- Gogate, P. R. and Pandit, A. B. (2004). A review of imperative technologies for wastewater treatment I: oxidation technologies at ambient conditions. *Advances in Environmental Research*, 8(3-4), 501-551.
- Guo, J., Xia, S., Wang, R. and Zhao, J. (2008). Study on membrane fouling of submerged membrane bioreactor in treating bathing wastewater. *Journal of Environmental Sciences*, 20(10), 1158-1167.
- Hach. (2008). *Water Analysis Handbook* (5th. ed.). U.S.A: Hach Company.
- Hai, F. I., Yamamoto, K., Nakajima, F. and Fukushi, K. (2012). Application of a GAC-coated hollow fiber module to couple enzymatic degradation of dye on membrane to whole cell biodegradation within a membrane bioreactor. *Journal of Membrane Science*, 389(0), 67-75.
- Hasan, D. u. B., Abdul Aziz, A. R. and Daud, W. M. A. W. (2012). Oxidative mineralisation of petroleum refinery effluent using Fenton-like process. *Chemical Engineering Research and Design*, 90(2), 298-307.
- Helen, W. (2005). Oil refineries: a review of their ecological impacts on the aquatic environment. *Estuarine, Coastal and Shelf Science*, 62(1-2), 131-140.
- Herney-Ramirez, J., Vicente, M. A. and Madeira, L. M. (2010). Heterogeneous photo-Fenton oxidation with pillared clay-based catalysts for wastewater treatment: A review. *Applied Catalysis B: Environmental*, 98(1-2), 10-26.

- Herrmann, J.-M. (1999). Heterogeneous photocatalysis: fundamentals and applications to the removal of various types of aqueous pollutants. *Catalysis Today*, 53(1), 115-129.
- Hiramoto, K., Ryuno, Y. and Kikugawa, K. (2002). Decomposition of N-nitrosamines, and concomitant release of nitric oxide by Fenton reagent under physiological conditions. *Mutation Research/Genetic Toxicology and Environmental Mutagenesis*, 520(1-2), 103-111.
- Huang, Y.-H., Huang, Y.-F., Chang, P.-S. and Chen, C.-Y. (2008). Comparative study of oxidation of dye-Reactive Black B by different advanced oxidation processes: Fenton, electro-Fenton and photo-Fenton. *Journal of Hazardous Materials*, 154(1-3), 655-662.
- Iurascu, B., Siminiceanu, I., Vione, D., Vicente, M. A. and Gil, A. (2009). Phenol degradation in water through a heterogeneous photo-Fenton process catalyzed by Fe-treated laponite. *Water Research*, 43(5), 1313-1322.
- Jantsch, T. G., Angelidaki, I., Schmidt, J. E., Brana de Hvidsten, B. E. and Ahring, B. K. (2002). Anaerobic biodegradation of spent sulphite liquor in a UASB reactor. *Bioresource Technology*, 84(1), 15-20.
- Javier Benitez, F., Acero, J. L. and Real, F. J. (2002). Degradation of carbofuran by using ozone, UV radiation and advanced oxidation processes. *Journal of Hazardous Materials*, 89(1), 51-65.
- Jefferson, B., Laine, A. L., Judd, S. J. and Stephenson, K. (2000). Membrane bioreactors and their role in wastewater reuse. *Water Sci. Technol.*, 41(1), 197-205.
- Jiang, T., Kennedy, M. D., van der Meer, W. G. J., Vanrolleghem, P. A. and Schippers, J. C. (2003). The role of blocking and cake filtration in MBR fouling. *Desalination*, 157(1-3), 335-343.
- Jiang, Y., Petrier, C. and Waite, T. D. (2006). Sonolysis of 4-chlorophenol in aqueous solution: Effects of substrate concentration, aqueous temperature and ultrasonic frequency. *Ultrasonics Sonochemistry*, 13(5), 415-422.
- Jou, C. J. G. and Huang, G.-C. (2003). A pilot study for oil refinery wastewater treatment using a fixed-film bioreactor. *Advances in Environmental Research*, 7(2), 463-469.
- Judd, S. (2006). *The MBR Book* (1st ed.): Elsevier.

- Khongnakorn, W., Wisniewski, C., Pottier, L. and Vachoud, L. (2007). Physical properties of activated sludge in a submerged membrane bioreactor and relation with membrane fouling. *Separation and Purification Technology*, 55(1), 125-131.
- Kim, K.-H. and Ihm, S.-K. (2011). Heterogeneous catalytic wet air oxidation of refractory organic pollutants in industrial wastewaters: A review. *Journal of Hazardous Materials*, 186(1), 16-34.
- Kitis, M., Adams, C. D. and Daigger, G. T. (1999). The effects of Fenton's reagent pretreatment on the biodegradability of nonionic surfactants. *Water Research*, 33(11), 2561-2568.
- Klamerth, N., Rizzo, L., Malato, S., Maldonado, M. I., Aguera, A. and Fernandez-Alba, A. R. (2009). Degradation of fifteen emerging contaminants at microg L(-1) initial concentrations by mild solar photo-Fenton in MWTP effluents. *Water Res.*
- Kolhatkar, A. and Sublette, K. (1996). Biotreatment of refinery spent sulfidic caustic by specialized cultures and acclimated activated sludge. *Applied Biochemistry and Biotechnology*, 57-58(1), 945-957.
- Kumfer, B., Felch, C., and Maugans, C., (2010). Wet air Oxidation Treatment of Spent Caustic in Petroleum Refineries. Proceedings of the 2010 *National Petroleum Refiner's Association Conference*. 21-23/3/2010.
- Kuyukina, M. S., Ivshina, I. B., Serebrennikova, M. K., Krivorutchko, A. B., Podorozhko, E. A. and Ivanov, R. V. (2009). Petroleum-contaminated water treatment in a fluidized-bed bioreactor with immobilized Rhodococcus cells. *International Biodeterioration & Biodegradation*, 63(4), 427-432.
- Kwon, B. G., Lee, D. S., Kang, N. and Yoon, J. (1999). Characteristics of p-chlorophenol oxidation by Fenton's reagent. *Water Research*, 33(9), 2110-2118.
- Le-Clech, P., Chen, V. and Fane, T. A. G. (2006). Fouling in membrane bioreactors used in wastewater treatment. *Journal of Membrane Science*, 284(1-2), 17-53.
- Le Clech, P., Jefferson, B., Chang, I. S. and Judd, S. J. (2003). Critical flux determination by the flux-step method in a submerged membrane bioreactor. *Journal of Membrane Science*, 227(1-2), 81-93.



- Li, H., Zhou, S., Sun, Y. and Lv, J. (2010). Application of response surface methodology to the advanced treatment of biologically stabilized landfill leachate using Fenton's reagent. *Waste Management*, 30(11), 2122-2129.
- Lin, H., Gao, W., Meng, F., Liao, B. Q., Leung, K. T. and Zhao, L. (2012). Membrane bioreactors for industrial wastewater treatment: A critical review. *Critical Reviews in Environmental Science and Technology*, 42(7), 677-740.
- Lin, S. H. and Ho, S. J. (1997). Treatment of high-strength industrial wastewater by wet air oxidation-A case study. *Waste Management*, 17(1), 71-78.
- Lin, S. H. and Kiang, C. D. (2003). Combined physical, chemical and biological treatments of wastewater containing organics from a semiconductor plant. *Journal of Hazardous Materials*, 97(1-3), 159-171.
- Lin, S. H., Lin, c. M. and Leu, H. G. (1999). Operating characteristics and kinetic studies of surfactant wastewater treatment by Fenton oxidation. *Water Research*, 33(7), 1735-1741.
- Lin, S. H. and Lo, C. C. (1997). Fenton process for treatment of desizing wastewater. *Water Research*, 31(8), 2050-2056.
- Liu, H.-L. and Chiou, Y.-R. (2005). Optimal decolorization efficiency of Reactive Red 239 by UV/TiO<sub>2</sub> photocatalytic process coupled with response surface methodology. *Chemical Engineering Journal*, 112(1-3), 173-179.
- Lodha, B. and Chaudhari, S. (2007). Optimization of Fenton-biological treatment scheme for the treatment of aqueous dye solutions. *Journal of Hazardous Materials*, 148(1-2), 459-466.
- Lohwacharin, J. and Annachhatre, A. P. (2010). Biological sulfide oxidation in an airlift bioreactor. *Bioresource Technology*, 101(7), 2114-2120.
- Lu, M. C., Lin, C. J., Liao, C. H., Ting, W. P. and Huang, R. Y. (2001) Influence of pH on the dewatering of activated sludge by Fenton's reagent. *Vol. 44* (pp. 327-332).
- Ma, F., Guo, J. b., Zhao, L. j., Chang, C. c. and Cui, D. (2009). Application of bioaugmentation to improve the activated sludge system into the contact oxidation system treating petrochemical wastewater. *Bioresource Technology*, 100(2), 597-602.
- Malato, S., Blanco, J. n., Alarcon, D. C., Maldonado, M. I., Fernandez-Ibanez, P. and Gernjak, W. (2007). Photocatalytic decontamination and disinfection of water with solar collectors. *Catalysis Today*, 122(1-2), 137-149.

- Malato, S., Blanco, J. n., Vidal, A. and Richter, C. (2002). Photocatalysis with solar energy at a pilot-plant scale: an overview. *Applied Catalysis B: Environmental*, 37(1), 1-15.
- Malaysia, D. o. E. (1979). Environmental Quality (Sewage and Industrial Effluents) Regulations 1979 *Third schedule Regulation 8(1), 8 (2) and 8 (3)*.
- Martin, M. A., Raposo, F., Borja, R. and Martin, A. (2002). Kinetic study of the anaerobic digestion of vinasse pretreated with ozone, ozone plus ultraviolet light, and ozone plus ultraviolet light in the presence of titanium dioxide. *Process Biochemistry*, 37(7), 699-706.
- Martins, R. C., Rossi, A. F. and Quinta-Ferreira, R. M. (2010). Fenton's oxidation process for phenolic wastewater remediation and biodegradability enhancement. *Journal of Hazardous Materials*, 180(1-3), 716-721.
- Mater, L., Rosa, E. V. C., Berto, J., Correa, A. X. R., Schwingel, P. R. and Radetski, C. M. (2007). A simple methodology to evaluate influence of  $H_2O_2$  and  $Fe^{2+}$  concentrations on the mineralization and biodegradability of organic compounds in water and soil contaminated with crude petroleum. *Journal of Hazardous Materials*, 149(2), 379-386.
- Mishra, V. S., Mahajani, V. V. and Joshi, J. B. (1995). Wet Air Oxidation. *Industrial & Engineering Chemistry Research*, 34(1), 2-48.
- Mohammed, T. A., Birima, A. H., Noor, M. J. M. M., Muyibi, S. A. and Idris, A. (2008). Evaluation of using membrane bioreactor for treating municipal wastewater at different operating conditions. *Desalination*, 221(1-3), 502-510.
- Moraes, J. E. R. F., Quina, F. H., Nascimento, C. U. A. O., Silva, D. N. and Chiavone-Filho, O. (2004). Treatment of Saline Wastewater Contaminated with Hydrocarbons by the Photo-Fenton Process. *Environmental Science & Technology*, 38(4), 1183-1187.
- Mrayyan, B. and Battikhi, M. N. (2005). Biodegradation of total organic carbons (TOC) in Jordanian petroleum sludge. *Journal of Hazardous Materials*, 120(1-3), 127-134.
- Mutamim, N. S. A., Noor, Z. Z., Hassan, M. A. A. and Olsson, G. (2012). Application of membrane bioreactor technology in treating high strength industrial wastewater: a performance review. *Desalination*, 305(0), 1-11.

- Mutamim, N. S. A., Noor, Z. Z., Hassan, M. A. A., Yuniarto, A. and Olsson, G. (2013). Membrane bioreactor: Applications and limitations in treating high strength industrial wastewater. *Chemical Engineering Journal*, 225(0), 109-119.
- N. L. Laoufi, D. T., F. Bentahar. (2008). The Degradation Of Phenol In Water Solution By  $\text{TiO}_2$  Photocatalysis In A Helical Reactor. *Global Nest Journal*, 10(3), 404-418.
- Najjar, W., Azabou, S., Sayadi, S. and Ghorbel, A. (2007). Catalytic wet peroxide photo-oxidation of phenolic olive oil mill wastewater contaminants: Part I. Reactivity of tyrosol over (Al-Fe) PILC. *Applied Catalysis B: Environmental*, 74(1-2), 11-18.
- Nesheiwat, F. K. and Swanson, A. G. (2000). Clean contaminated sites using Fenton's reagent. *Chemical Engineering Progress*, 96(4), 61-66.
- Neyens, E. and Baeyens, J. (2003). A review of classic Fenton's peroxidation as an advanced oxidation technique. *Journal of Hazardous Materials*, 98(1-3), 33-50.
- Oller, I., Malato, S. and Sanchez-Perez, J. A. (2011). Combination of Advanced Oxidation Processes and biological treatments for wastewater decontamination- A review. *Science of The Total Environment*, 409(20), 4141-4166.
- Olmos, A., Olguin, P., Fajardo, C., Razo, E. and Monroy, O. (2004). Physicochemical Characterization of Spent Caustic from the OXIMER Process and Sour Waters from Mexican Oil Refineries. *Energy & Fuels*, 18(2), 302-304.
- Pardeshi, S. K. and Patil, A. B. (2008). A simple route for photocatalytic degradation of phenol in aqueous zinc oxide suspension using solar energy. *Solar Energy*, 82(8), 700-705.
- Pera-Titus, M., Garcia-Molina, V., Banos, M. A., Gimenez, J. and Esplugas, S. (2004). Degradation of chlorophenols by means of advanced oxidation processes: a general review. *Applied Catalysis B: Environmental*, 47(4), 219-256.
- Pignatello, J. J. (1992). Dark and photoassisted iron(3+)-catalyzed degradation of chlorophenoxy herbicides by hydrogen peroxide. *Environmental Science & Technology*, 26(5), 944-951.

- Pignatello, J. J., Oliveros, E. and MacKay, A. (2007). Erratum: Advanced oxidation processes for organic contaminant destruction based on the fenton reaction and related chemistry (Critical Reviews in Environmental Science and Technology (2006) 36, (1-84)). *Critical Reviews in Environmental Science and Technology*, 37(3), 273-275.
- Poulton, S. W., Krom, M. D., Rijn, J. V. and Raiswell, R. (2002). The use of hydrous iron (III) oxides for the removal of hydrogen sulphide in aqueous systems. *Water Research*, 36(4), 825-834.
- Pulgarin, C., Invernizzi, M., Parra, S., Sarria, V., Polania, R. and Pöhringer, P. (1999). Strategy for the coupling of photochemical and biological flow reactors useful in mineralization of biorecalcitrant industrial pollutants. *Catalysis Today*, 54(2-3), 341-352.
- Puspitasari, V., Granville, A., Le-Clech, P. and Chen, V. (2010). Cleaning and ageing effect of sodium hypochlorite on polyvinylidene fluoride (PVDF) membrane. *Separation and Purification Technology*, 72(3), 301-308.
- Rahman, M. M. and Al-Malack, M. H. (2006). Performance of a crossflow membrane bioreactor (CF-MBR) when treating refinery wastewater. *Desalination*, 191(1-3), 16-26.
- Ray, A. K. (2009). Photocatalytic Reactor Configurations for Water Purification: Experimentation and Modeling. In I. d. L. Hugo & R. Benito Serrano (Eds.), *Advances in Chemical Engineering* (Vol. Volume 36, pp. 145-184): Academic Press.
- Rivas, F. J., Beltran, F. J., Frades, J. s. and Buxeda, P. (2001). Oxidation of p-hydroxybenzoic acid by Fenton's reagent. *Water Research*, 35(2), 387-396.
- Rodrigues, C. S. D., Meadeira, L. M. and BOaventura, R. A. R. (2009). Treatment of Textile Effluent by Chemical (Fenton's Reagent) and Biological(Sequencing Batch Reactor). *Journal of Hazardous Materials*, (172), 1551-1559.
- Rosenberger, S., Kruger, U., Witzig, R., Manz, W., Szewzyk, U. and Kraume, M. (2002). Performance of a bioreactor with submerged membranes for aerobic treatment of municipal waste water. *Water Research*, 36(2), 413-420.
- Saien, J. and Nejati, H. (2007). Enhanced photocatalytic degradation of pollutants in petroleum refinery wastewater under mild conditions. *Journal of Hazardous Materials*, 148(1-2), 491-495.

- Santos, F. V., Azevedo, E. B., Sant'Anna Jr, G. L. and Dezotti, M. (2006). Photocatalysis as a tertiary treatment for petroleum refinery wastewaters. *Brazilian Journal of Chemical Engineering*, 23(4), 451-460.
- Sedlak, D. L. and Andren, A. W. (1991). Oxidation of chlorobenzene with Fenton's reagent. *Environmental Science and Technology*, 25(4), 777-782.
- Senogles, P. J., Scott, J. A., Shaw, G. and Stratton, H. (2001). Photocatalytic Degradation of the Cyanotoxin Cylindrospermopsin, using Titanium Dioxide and UV Irradiation. *Water Research*, 35(5), 1245-1255.
- Shahrezaei, F., Mansouri, Y., Zinatizadeh, A. A. L. and Akhbari, A. (2012). Process modeling and kinetic evaluation of petroleum refinery wastewater treatment in a photocatalytic reactor using TiO<sub>2</sub> nanoparticles. *Powder Technology*, 221(0), 203-212.
- Sheu, S. H. and Weng, H. S. (2001). Treatment of Olefin Plant Spent Caustic by Combination of Neutralization and Fenton Reaction. *Water Research*, 35(8), 2017-2021.
- Sipma, J., Svitelskaya, A., van der Mark, B., Hulshoff Pol, L. W., Lettinga, G., and Buisman, C. J. N. (2004). Potentials of biological oxidation processes for the treatment of spent sulfidic caustics containing thiols. *Water Research*, 38(20), 4331-4340.
- Stasinakis, A. S. (2008). Use of Selected Advanced Oxidation Processes (AOPs) for Wastewater Treatment - A Mini Review. *Global Nest Journal*, 10, 376-385.
- Stephenson, K., Judd, S., Jefferson, B. and Brindle, K. (1996). *Membrane Bioreactors for Wastewater Treatment*. London, United Kingdom: IWA Publishing.
- Stepnowski, P., Siedlecka, E. M., Behrend, P. and Jastorff, B. (2002). Enhanced photo-degradation of contaminants in petroleum refinery wastewater. *Water Research*, 36(9), 2167-2172.
- Sum, O. S. N., Feng, J., Hub, X. and Yue, P. L. (2005). Photo-assisted fenton mineralization of an azo-dye acid black 1 using a modified laponite clay-based Fe nanocomposite as a heterogeneous catalyst. *Topics in Catalysis*, 33(1), 233-242.
- Tang, W. Z. and Huang, C. P. (1996). 2,4-Dichlorophenol Oxidation Kinetics by Fenton's Reagent. *Environmental Technology*, 17(12), 1371-1378.

- Tang, W. Z. and Tassos, S. (1997). Oxidation kinetics and mechanisms of trihalomethanes by Fenton's reagent. *Water Research*, 31(5), 1117-1125.
- Tansens, P., Rodal, A. T., Machado, C. M. M. and Soares, H. M. V. M. (2011). Recycling of aluminum and caustic soda solution from waste effluents generated during the cleaning of the extruder matrixes of the aluminum industry. *Journal of Hazardous Materials*, 187(1-3), 459-465.
- Tizaoui, C., Bouselmi, L., Mansouri, L. and Ghrabi, A. (2007). Landfill leachate treatment with ozone and ozone/hydrogen peroxide systems. *Journal of Hazardous Materials*, 140(1-2), 316-324.
- Torrades, F., Garcia-Montano, J., Garcia-Hortal, J. A., Domenech, X. and Jose, P. (2004). Decolorization and mineralization of commercial reactive dyes under solar light assisted photo-Fenton conditions. *Solar Energy*, 77(5), 573-581.
- Vaiopoulou, E., Melidis, P. and Aivasidis, A. (2005). Sulfide removal in wastewater from petrochemical industries by autotrophic denitrification. *Water Research*, 39(17), 4101-4109.
- Broeck, V. d. R., Dierdonck, V. J., Nijskens, P., Dotremont, C., Krzeminski, P. and Graaf, v. d. J. H. J. M. (2012). The influence of solids retention time on activated sludge bioflocculation and membrane fouling in a membrane bioreactor (MBR). *Journal of Membrane Science*, 401-402(0), 48-55.
- Villain, M. and Marrot, B. (2013). Influence of sludge retention time at constant food to microorganisms ratio on membrane bioreactor performances under stable and unstable state conditions. *Bioresource Technology*, 128(0), 134-144.
- Vogel, F. D. R., Blanchard, J. L. D., Marrone, P. A., Rice, S. F., Webley, P. A., Peters, W. A., (2005). Critical review of kinetic data for the oxidation of methanol in supercritical water. *The Journal of Supercritical Fluids*, 34(3), 249-286.
- Wang, P., Wang, Z., Wu, Z. and Mai, S. (2011). Fouling behaviours of two membranes in a submerged membrane bioreactor for municipal wastewater treatment. *Journal of Membrane Science*, 382(1-2), 60-69.
- Wang, Z., Wu, Z., Yin, X. and Tian, L. (2008). Membrane fouling in a submerged membrane bioreactor (MBR) under sub-critical flux operation: Membrane foulant and gel layer characterization. *Journal of Membrane Science*, 325(1), 238-244.

- Wu, Y., Zhou, S., Qin, F., Ye, X. and Zheng, K. (2010). Modeling physical and oxidative removal properties of Fenton process for treatment of landfill leachate using response surface methodology (RSM). *Journal of Hazardous Materials*, 180(1-3), 456-465.
- Xing, C. H., Tardieu, E., Qian, Y. and Wen, X. H. (2000). Ultrafiltration membrane bioreactor for urban wastewater reclamation. *Journal of Membrane Science*, 177(1-2), 73-82.
- Xu, J. and Gao, C. (2010). Study of critical flux in ultrafiltration of seawater: New measurement and sub- and super-critical flux operations. *Chemical Engineering Journal*, 165(1), 102-110.
- Yan, L., Ma, H., Wang, B., Wang, Y. and Chen, Y. (2011). Electrochemical treatment of petroleum refinery wastewater with three-dimensional multi-phase electrode. *Desalination*, 276(1-3), 397-402.
- Yigit, N. O., Uzal, N., Koseoglu, H., Harman, I., Yukseler, H. and Yetis, U. (2009). Treatment of a denim producing textile industry wastewater using pilot-scale membrane bioreactor. *Desalination*, 240(1-3), 143-150.
- Yoon, J., Lee, Y. and Kim, S. (2001) Investigation of the reaction pathway of OH radicals produced by Fenton oxidation in the conditions of wastewater treatment. *Vol. 44* (pp. 15-21).
- Zainal-Abideen, M., Aris, A., Yusof, F., Abdul-Majid, Z., Selamat, A. and Omar, S. I. (2012). Optimizing the coagulation process in a drinking water treatment plant - Comparison between traditional and statistical experimental design jar tests. *Water Science and Technology*, 65(3), 496-503.
- Zhao, W., Mou, Q., Zhang, X., Shi, J., Sun, S. and Zhao, C. (2013). Preparation and characterization of sulfonated polyethersulfone membranes by a facile approach. *European Polymer Journal*, 49(3), 738-751.
- Zondervan, E. and Roffel, B. (2007). Evaluation of different cleaning agents used for cleaning ultra filtration membranes fouled by surface water. *Journal of Membrane Science*, 304(1-2), 40-49.